

	<b>Positive Ion</b>	<b>Negative Ion</b>	<b>Formula</b>	<b>Compound Name</b>
1				cobalt (II) chloride
2	$\text{Cr}^{2+}$	$\text{O}^{2-}$		
3			$\text{AuCl}$	
4	$\text{Pb}^{2+}$	$\text{F}^-$		
5	$\text{Cu}^+$	$\text{N}^{3+}$		
6			$\text{SnS}_2$	
7				copper (II) sulphide
8	$\text{Mn}^{3+}$	$\text{O}^{2-}$		
9				gold (III) sulphide
10			$\text{Fe}_2\text{S}_3$	
11				cobalt (III) oxide
12				manganese (III) nitride
13			$\text{UF}_6$	
14			$\text{Re}_3\text{As}_7$	
15				niobium (V) iodide
16				copper (I) telluride
17			$\text{TlCl}$	
18				molybdenum (III) sulphide
19	$\text{Rh}^{4+}$	$\text{Br}^-$		
20	$\text{Sb}^{5+}$	$\text{S}^{2-}$		
21	$\text{Sn}^{2+}$	$\text{F}^-$		
22				bismuth (V) phosphide
23				vanadium (V) oxide
24	$\text{Ti}^{4+}$	$\text{O}^{2-}$		
25			$\text{NiBr}_2$	